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Rainfall and throughfall chemistry in the Atlantic Forest: a comparison between urban and natural areas (São Paulo State, Brazil)

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Abstract

Two areas in the Atlantic Forest (São Paulo State, Brazil), with contrasting environments in respect of human occupation, were monitored from 1999 to 2001. One area named PEFI (23°38'08''–23°40'18'' S and 46°36'48''–46°38'00'' W) at an altitude of 798 m a.s.l., 526.4 ha in area and about 50 km from the sea, lies in a State Park within the largest metropolis of South America – São Paulo. The other area, named CUNHA (between 23°13'18'' and 23°16'10'' S and 45°02'53'' and 45°05'15'' W) about 1050 m a.s.l. with an area of 2854 ha and about 15 km from the sea, is also within a State Park in the Atlantic Forest, but is surrounded by rural areas and small villages. For each area, the rainfall and throughfall chemistry were examined and pH and Na⁺, K⁺, Mg²⁺, Ca²⁺, NH₄⁺, Cl⁻, NO₃⁻ and SO₄²⁻ as well as trace metals were determined. Compared with PEFI, CUNHA is characterised by low chemical fluxes and the largest differences are for the ions such as Ca²⁺, H⁺, NO₃⁻ and SO₄²⁻ which are mainly anthropogenic in origin. Differences in throughfall chemical fluxes are linked to the nutritional status of the trees.

Keywords: Atlantic Forest, hydrochemistry, nutrient cycles, rainfall, throughfall, water chemistry

Introduction

In the past, the Atlantic Forest comprised 15% of Brazilian territory, covering as it did almost the whole of the Brazilian coastal region from 25°S (State of Santa Catarina) to 5°S (State of Rio Grande do Norte). Since the European colonisation of Brazil in the 16th century, the Atlantic forest has been exploited so that, nowadays, much of its original area is occupied by farms as well as urban and industrial centres, including the major urban centres of South America such as the Metropolitan Region of São Paulo city (State of São Paulo) which is the fourth largest urban centre in the world with 17 million inhabitants. At present, only about 15% of the original Atlantic Forest in the State of São Paulo is preserved as the Biosphere Reserve. Although protected legally, these areas are still under anthropogenic stress from the presence, nearby, of urban and industrial centres as well

as from exposure to the deposition of pollutants transported through long distances from such centres.

This paper presents the results of a hydrogeochemical study of two key locations within the Atlantic Forest, mainly to characterise the evolution of the water chemistry of the rainfall and throughfall over time and to examine regional differences in the solute depositions from the atmosphere to the soil and vegetative canopy. The study was undertaken in the State of São Paulo, Brazil, at sites with contrasting pollution climates; the one is polluted, while the other is nearly pristine. The changes occurring are linked to the nutritional status of the vegetation; temporal and spatial differences in the concentrations of solute transferred from the atmosphere to the vegetation cover and thence to the soil surface are examined. Here, a hypothesis is tested: forested areas constantly exposed to pollutants are prone to

degradation of their vegetation. A major environmental issue for Brazil is protection of the functioning of the ecosystem as well as of the biological resource of the Atlantic Forest.

Study areas

The urban forest studied is in the 'Parque Estadual das Fontes do Ipiranga' (PEFI) between parallels 23°38'08" and 23°40'18"S and meridians 46°36'48" and 46°38'00"W at a mean altitude above sea level of 798 m (Fernandes *et al.*, 2002) and 50 km from the sea shore. The PEFI is a biological reserve 526.4 ha in area within the metropolitan region of São Paulo city in the south-eastern region of São Paulo City County. PEFI includes an area of water source preservation and is a reserve of biodiversity, a natural laboratory for basic and applied research, a controlling factor for micro-climate, a protection for the water table, a natural laboratory for educational programmes and an important recreation area for the community (Bicudo *et al.*, 2002). Its climate is

influenced by extra-tropical (e.g. frontal systems) and tropical (e.g. squall lines and convergence zones) systems. The mean annual rainfall (1933–1999) is 1368 mm, with a dry period from May to August. The mean temperature for the coldest month, July, is 15°C and for January and February, the two hottest months, 21–22°C (Santos and Funari, 2002). It is on crystalline rocks with gneisses and secondary micaschists, overlain by clay, sand and gravels of the São Paulo Formation. The topography is undulating, with heights of 770 m to 825 m, being part of the central ridge hills of São Paulo city (Almeida, 1974).

The 'CUNHA' is at 1050 m a.s.l., 15 km from the sea between 23°13'28" and 23°16'10" S and 45°02'53" and 45°05'15" W with an area of 2854 ha.

It hosts the *Walter Emmerich* Forest Hydrology Laboratory of the Institute of Forestry of the State of São Paulo. The mean maximum and minimum temperatures are 26°C and 16°C, respectively. The region can be classed as 'mist-tropical' and mean annual rainfall is 1982 mm (1982–2000) with the main rainy period from September until April. The geology of the area belongs to the Brazilian crystalline complex of Pre-Cambrian with granites, gneisses, crystalline schist and diabase rocks of Mesozoic age. The topography is undulating, with low, well rounded hills, valleys with slopes of 10 to 20 degrees and a low drainage density with heights from 1050 to 1170 m a.s.l. (Arcova *et al.*, 1993). Figure 1 shows the relative position of the two areas studied.

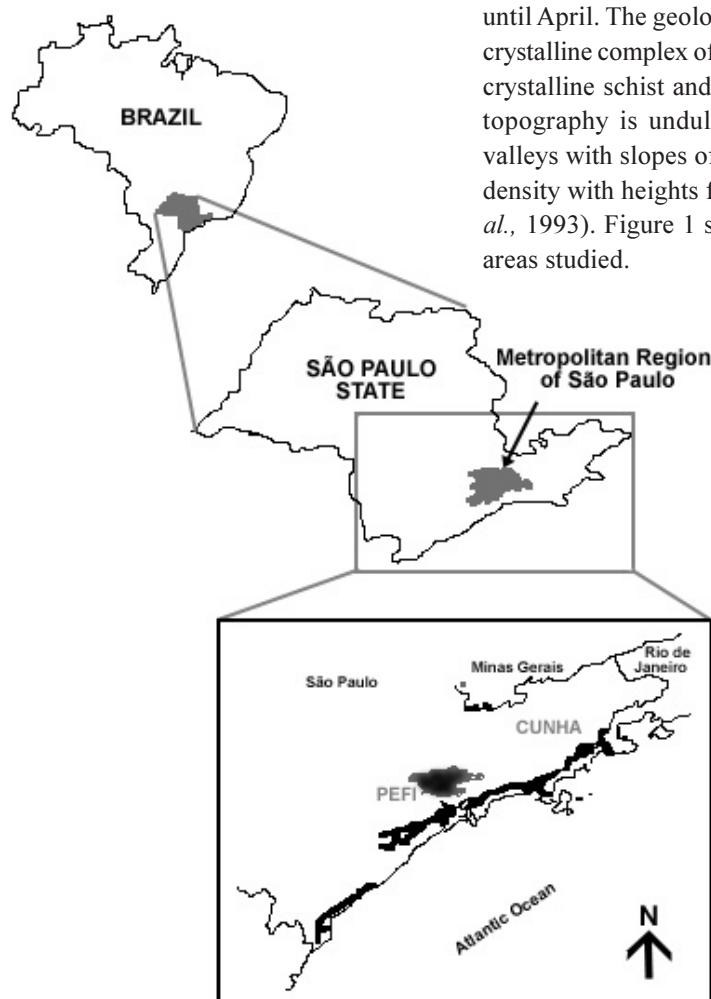


Fig. 1. Schematic map showing the relative positions of the São Paulo city metropolitan region, PEFI, and CUNHA.

Methods

SAMPLING PROCEDURE

Rainfall and throughfall water for two wet and two dry periods were sampled from June 1999 to December 2001 in the area of the urban forest (PEFI) (70 samples) and from April 2000 to March 2002 in the control forest (CUNHA) (75 samples). Here, the dry and wet periods refer to the driest (May to August) and the wettest (November to February) months. The water samples were accumulated weekly but, if there was insufficient water after 15 days' exposure, the collecting funnel and bottles were substituted with clean ones to minimise dry deposition. The rainfall bulk collector is a polyethylene funnel 167.4 cm² in collecting area, covered with a nylon mesh of 5 mm to avoid contamination by insects, branches and leaves, coupled to a 5-L polyethylene bottle. A wet-only collector (MTX-Italy) was deployed simultaneously and the two sets of measurements were used to estimate the effect of dry deposition on the rainfall chemistry.

Throughfall was sampled with ten bulk collectors in a 6 × 6 matrix of 36 sampling points 5 m apart in a 30 × 30 m plot. Each of the 36 points was chosen to avoid trees and understorey vegetation and a 1.5m stake was deployed and identified with a number from 1 to 36. The top of each of the 36 stakes enabled a bulk collector to be installed or removed easily. For each sampling interval, the collectors were sited at ten different positions determined by generation of ten random numbers between 1 and 36. The volume of catch was determined gravimetrically. The ten bulk throughfall samples were mixed in proportion to the sample volumes into one composite sample for chemical analysis (Forti *et al.*, 1997; 2000a).

ANALYTICAL PROCEDURE

The waters were vacuum filtered with pre-washed membrane Millipore filters of 0.22 µm pore diameter to allow the determination of the dissolved fraction and to eliminate micro-organisms and fungi spores that might contaminate and degrade the sample. Three sub-samples were taken for chemical analysis: one stored in a high density polyethylene bottle without preservation for anion analysis, one acidified with supra-pure Nitric Acid (1% in volume) for the aluminium and transition metal analysis and one acidified with hydrochloric acid (0.1% in volume) for cation analysis. After the appropriate preservation, all samples were stored at 4°C to avoid sample degradation.

Liquid Ion Chromatography (DIONEX DX500) was employed for the chemical analysis. For the major cations a CS12 analytical column was used with sulphuric acid as

eluent. For the anions, an AS4A analytical column was used. Conductivity detection was used for major cations and anions. For the trace metals (Fe III, CuII, Zn II and Mn II) a UV/Visible detector was used with a CS5A analytical column with post-column derivation with 4-(2-piridilasol) resorcinol and pyridine dicarboxylic acid as eluent, with detection at a wavelength of 530 nm; for aluminium the column was the same and the eluent was sulphuric acid/ammonium sulphate, with post-column reaction involving pyrocatecol violet and detection at a wavelength of 530 nm. The accuracy was checked against certified reference material samples (TM28 and TM 26.2, trace metal fortified water, National Water Research Institute, Ontario Canada, 1999), standards from the United Kingdom Centre for Ecology and Hydrology Hydrochemistry Laboratory (Robson and Neal, 1997) and CRM 409 No. 629 major compounds in simulated rainwater from the Community Bureau of Reference – BCR (Commission of the European Communities). The precision and accuracy of the analysis of the different chemical species in the samples were, respectively, 3% and 7% for Na⁺, 3% and 9% for K⁺, 8% and 11% for Mg²⁺, 8% and 9% for Ca²⁺, 10% and 11% for NH₄⁺, 3% and 6% for Cl⁻, 3.5% and 10% for NO₃⁻, 8% and 8% for SO₄²⁻, 5% and 7% for Al III, 11% and 8% for Cu II, 11% and 15% for Fe III, 4% and 7% for Mn II and 7% and 11% for Zn II.

PRINCIPAL COMPONENT ANALYSIS (PCA)

Principal component analysis (PCA) was used to reduce the variables to fewer linear combinations of the original variables (Johnson and Wichern, 1988). Broken-stick eigenvalues were used to define the number of interpretable axes (Jackson, 1993). Before computation, variables were log transformed to approximate more closely the linear relationships assumed in PCA. For data processing and plotting, FITOPAC (Shepherd, 1996) and PC-ORD version 3.0 for Windows (McCune and Mefford, 1999) packages were used.

Results and discussion

COMPARISON BETWEEN WET-ONLY AND BULK COLLECTORS

A test of the null hypothesis was applied to the wet-only and bulk rainwater samples, with a probability confidence interval of 95% and significance level of 5%. The results for CUNHA indicate that, except for NO₃⁻, no differences in the composition of the rainwater for all the chemical species of the two sampling groups were statistically

significant; this finding is in accord with studies in the Amazonian region (Forti *et al.*, 2000b). Correspondingly, the results for PEFI showed that the weighted mean concentration values for K^+ and Cl^- were significantly higher in the samples collected in the bulk collector: dry deposition over the bulk collector was significant during the collection period but the procedure to minimise dry deposition was inefficient for these ions.

Because of the lack of major differences between wet-only and bulk measurements, the study hereafter has been based on the concentrations determined from the samples collected with the bulk collector because one of the objectives of this work is the quantification of the solutes deposited on each ecosystem. The statistical analysis has concluded that rainwater samples accumulated in bulk collectors do not overestimate the concentrations of the chemical species in natural areas or those exposed to low anthropogenic influences. However, the collectors were not exposed for periods longer than seven days.

RAINFALL CHEMISTRY

Precipitation was measured monthly at the meteorological stations at each site; for the period of the study, these values and the estimates from volume measurements (rainfall and throughfall) are given in Figs. 2a and 2b.

Figures 3 (a, b and c) and 4 (a, b and c) present the mean monthly variations in rainwater chemistry for PEFI and CUNHA, respectively. The highest concentrations occur during periods of low rainfall at both sites. However, at PEFI, the concentrations are significantly higher during the driest months (May to August). For CUNHA, concentrations decrease slightly in the rainiest months (November to February). The concentrations vary over time at PEFI which underlines the importance of pollutant scavenging in the atmosphere.

Mean concentrations are calculated as the volume weighted mean (VWM); in Table 1, values for both forests are presented together with maxima and minima for the

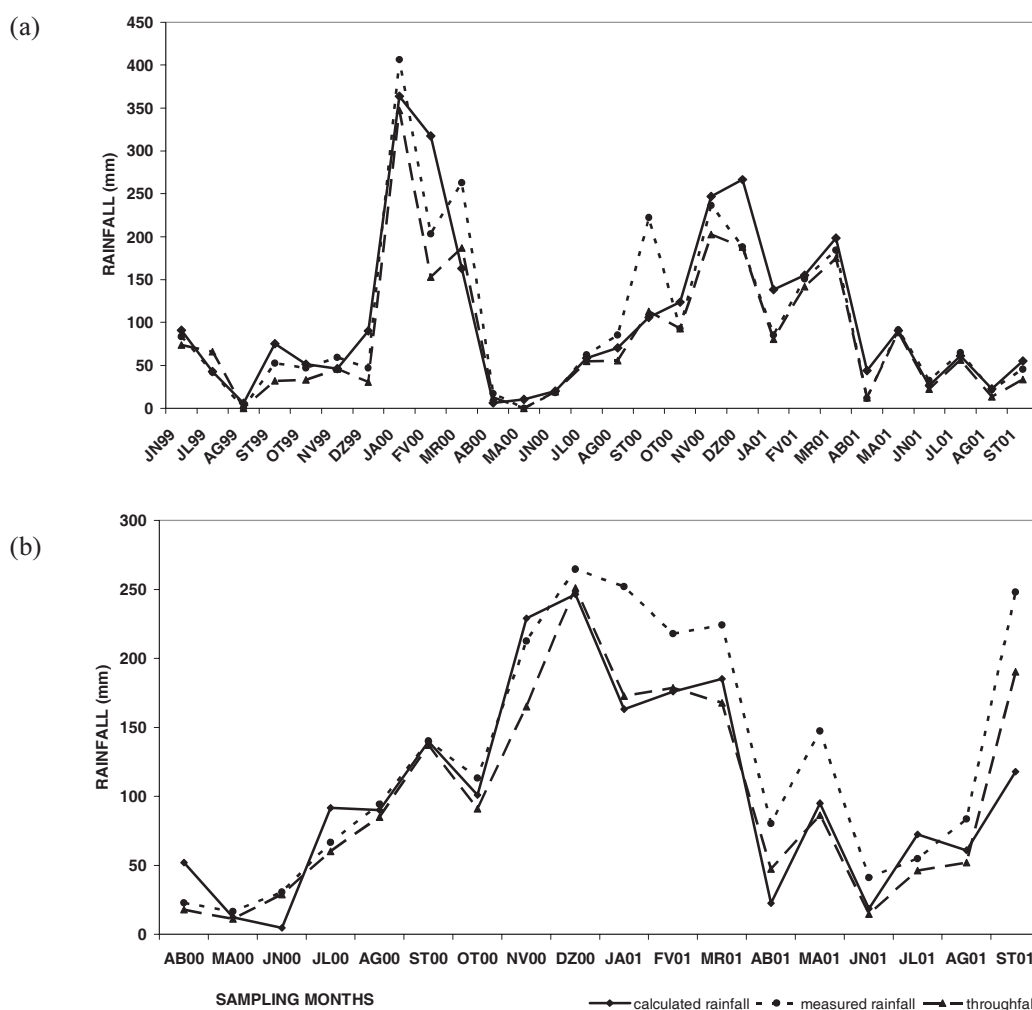


Fig. 2 Monthly mean rainfall in mm at the meteorological station and rainfall and throughfall values calculated from the measured volume for the period studied for (a) the PEFI area and (b) for the CUNHA area

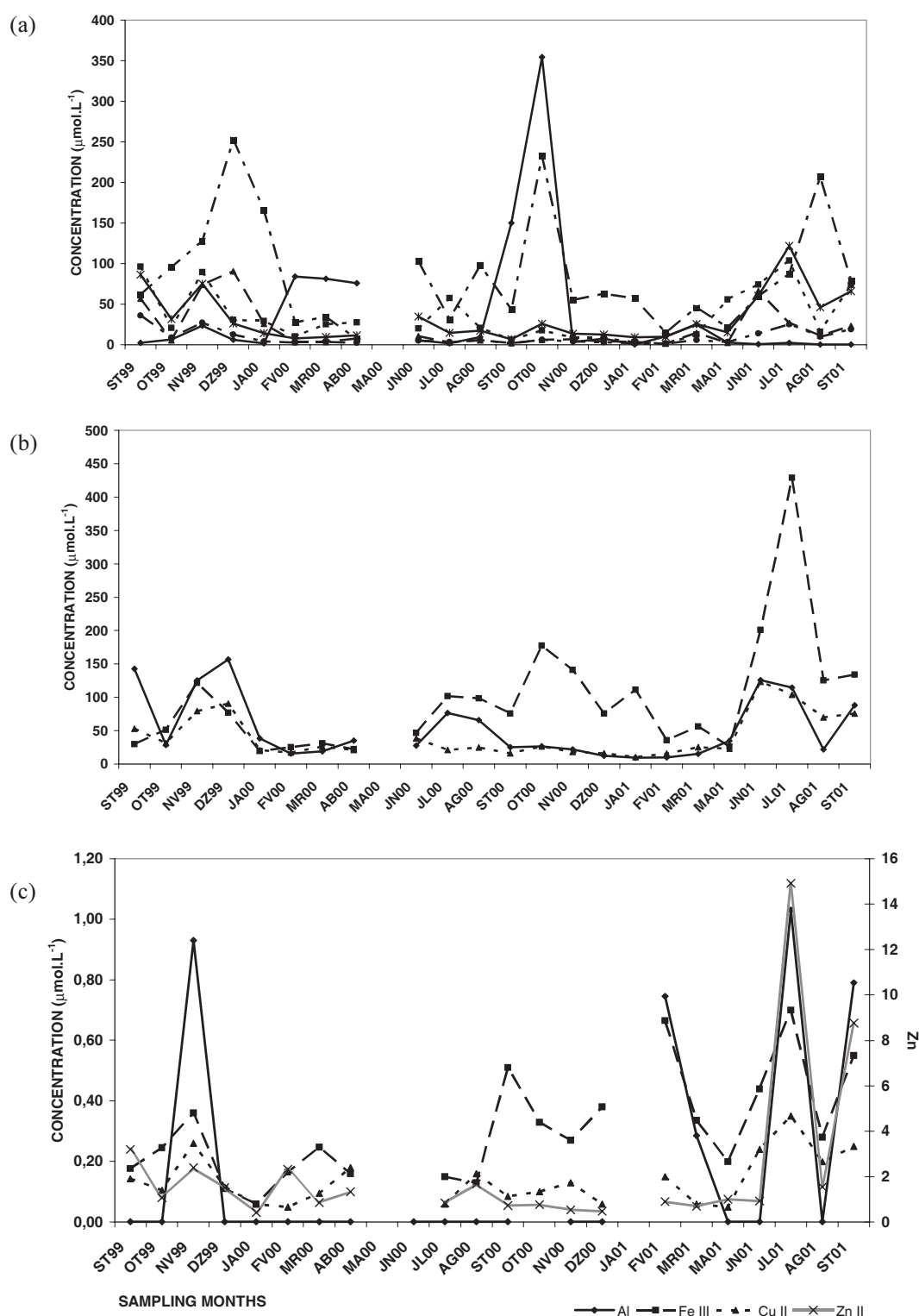


Fig. 3. Rainfall monthly mean concentration variations in different chemicals at PEFI: (a) major cations (b) major anions (c) transition metals.

whole sampling period as well as for the four rainiest and driest months. Clearly, concentrations vary to a large extent.

For PEFI, the rainwater is acidic with a mean pH of 4.3 and there is no significant statistical difference (95% of

confidence interval at 5% of confidence level) between the dry and rainy periods. The mean pH is about 5.4 for CUNHA, without significant seasonal differences. Trace metal concentrations for rainfall at CUNHA are generally

Table 1. Rainfall chemistry in $\mu\text{Eq.L}^{-1}$ for the whole period and the driest and wettest months obtained for PEFI and CUNHA. P: rainfall, mm; N: sample number; MAX: maximum values; MIN: minimum values, MP: volume weighted mean.

		PEFI			CUNHA		
		Total	Driest	Wettest	Total	Driest	Wettest
P	(mm)	3143	516.7	1840	3722	669.1	2006
H ⁺	N	62	26	36	50	23	27
	MAX	368	21.5	119	14.01	6.39	7.47
	MIN $\mu\text{Eq.L}^{-1}$	0.01	0.01	0.45	0.13	0.52	0.36
	MP	47.2	9.24	34.4	4.14	3.35	3.90
Na ⁺	MAX	233	233	33.3	41.5	41.5	14.3
	MIN $\mu\text{Eq.L}^{-1}$	2.72	3.41	2.72	1.70	4.26	1.70
	MP	15.5	31.8	11.6	8.33	12.4	5.19
K ⁺	MAX	155	155	52.6	31.3	7.61	22.5
	MIN $\mu\text{Eq.L}^{-1}$	1.56	1.79	1.56	0.58	0.58	1.12
	MP	11.0	12.4	12.9	8.61	2.39	9.69
Mg ²⁺	MAX	90.2	90.2	10.2	5.47	4.11	5.47
	MIN $\mu\text{Eq.L}^{-1}$	1.64	2.26	1.64	0.32	0.51	0.32
	MP	4.38	7.09	3.40	2.29	1.08	2.53
Ca ²⁺	MAX	214	214	24.4	4.24	3.69	4.24
	MIN $\mu\text{Eq.L}^{-1}$	5.88	9.60	5.88	0.00	0.00	0.54
	MP	14.7	24.6	10.7	1.89	1.22	2.01
NH ₄ ⁺	MAX	268	103	161	255	19.6	255
	MIN $\mu\text{Eq.L}^{-1}$	2.83	8.04	7.26	0.79	0.79	2.17
	MP	59.9	39.7	57.2	38.0	8.71	57.6
Cl ⁻	MAX	345	345	58.5	106	35.5	11.2
	MIN $\mu\text{Eq.L}^{-1}$	5.12	5.12	9.60	4.14	5.89	7.50
	MP	27.2	52.3	22.3	14.4	10.9	5.18
NO ₃ ⁻	MAX	427	427	135	33.1	22.2	12.1
	MIN $\mu\text{Eq.L}^{-1}$	5.83	5.83	13.6	0.28	3.43	0.50
	MP	65.2	113	49.0	8.00	9.69	5.53
SO ₄ ²⁻	MAX	133	133	47.2	11.5	7.64	7.63
	MIN $\mu\text{Eq.L}^{-1}$	6.43	13.8	6.43	1.78	1.78	2.07
	MP	23.7	29.5	21.4	4.16	3.80	3.69
Al III	N	43	16	27	20	13	7
	MAX	1.58	1.03	0.62	10.1	1.23	10.1
	MIN $\mu\text{Eq.L}^{-1}$	0.00	0.001	0.001	0.00	0.00	0.02
	MP	0.17	0.13	0.07	2.93	1.59	3.74
Fe III	MAX	0.87	0.70	0.47	0.31	0.05	0.31
	MIN $\mu\text{Eq.L}^{-1}$	0.06	0.15	0.06	0.00	0.00	0.00
	MP	0.27	0.24	0.21	0.04	0.02	0.05
Cu II	MAX	0.35	0.35	0.15	0.44	0.05	0.44
	MIN $\mu\text{Eq.L}^{-1}$	0.04	0.05	0.04	0.00	0.00	0.00
	MP	0.09	0.12	0.07	0.06	0.01	0.10
Zn II	MAX	16.0	14.9	3.30	2.06	2.01	1.20
	MIN $\mu\text{Eq.L}^{-1}$	0.38	0.38	0.40	0.00	0.14	0.00
	MP	2.18	2.72	1.00	0.24	0.36	0.23
Mn II	MAX	1.69	1.69	0.26	0.18	0.18	0.12
	MIN $\mu\text{Eq.L}^{-1}$	0.07	0.07	0.07	0.00	0.00	0.00
	MP	0.19	0.38	0.12	0.04	0.05	0.03

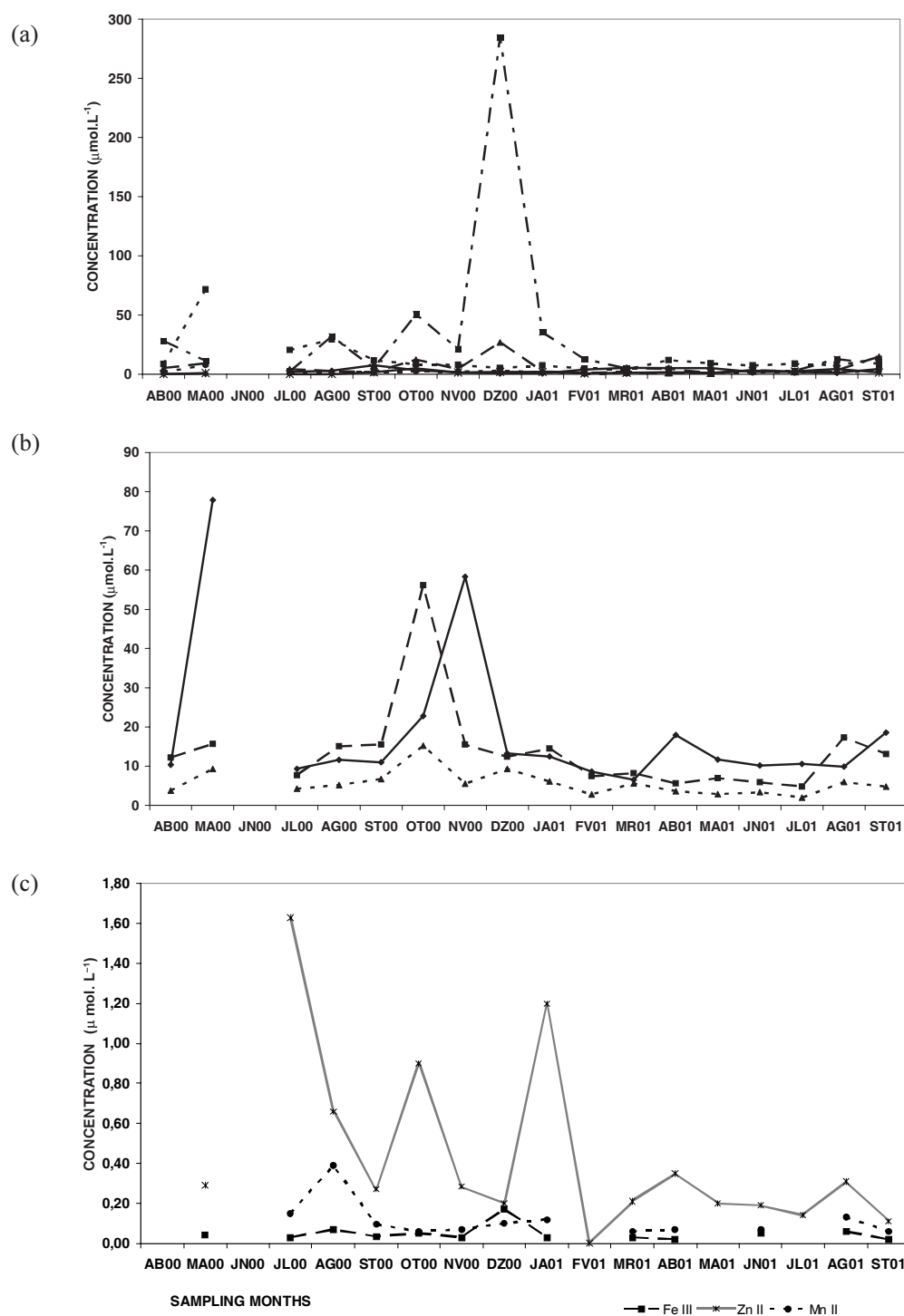


Fig. 4. Rainfall monthly mean concentration variations in different chemicals at CUNHA: (a) major cations (b) major anions (c) transition metals.

close to the detection limits, except for Zn II and Mn II, while for PEFI the concentrations are within the confidence intervals of the analytical method.

Statistical evaluation (to 95% probability) showed a difference between the wet and dry season for the rainfall but not for the mean values of all the chemical determinands at both sites.

THROUGHFALL CHEMISTRY

Figures 5 (a, b and c) and 6 (a, b and c) show mean monthly variations (VWM) for each chemical species in throughfall for PEFI and CUNHA, respectively. In both regions, the concentrations for all chemical species are significantly higher in the driest than in the wettest period, because the vegetation is an efficient filter for the airborne particulates

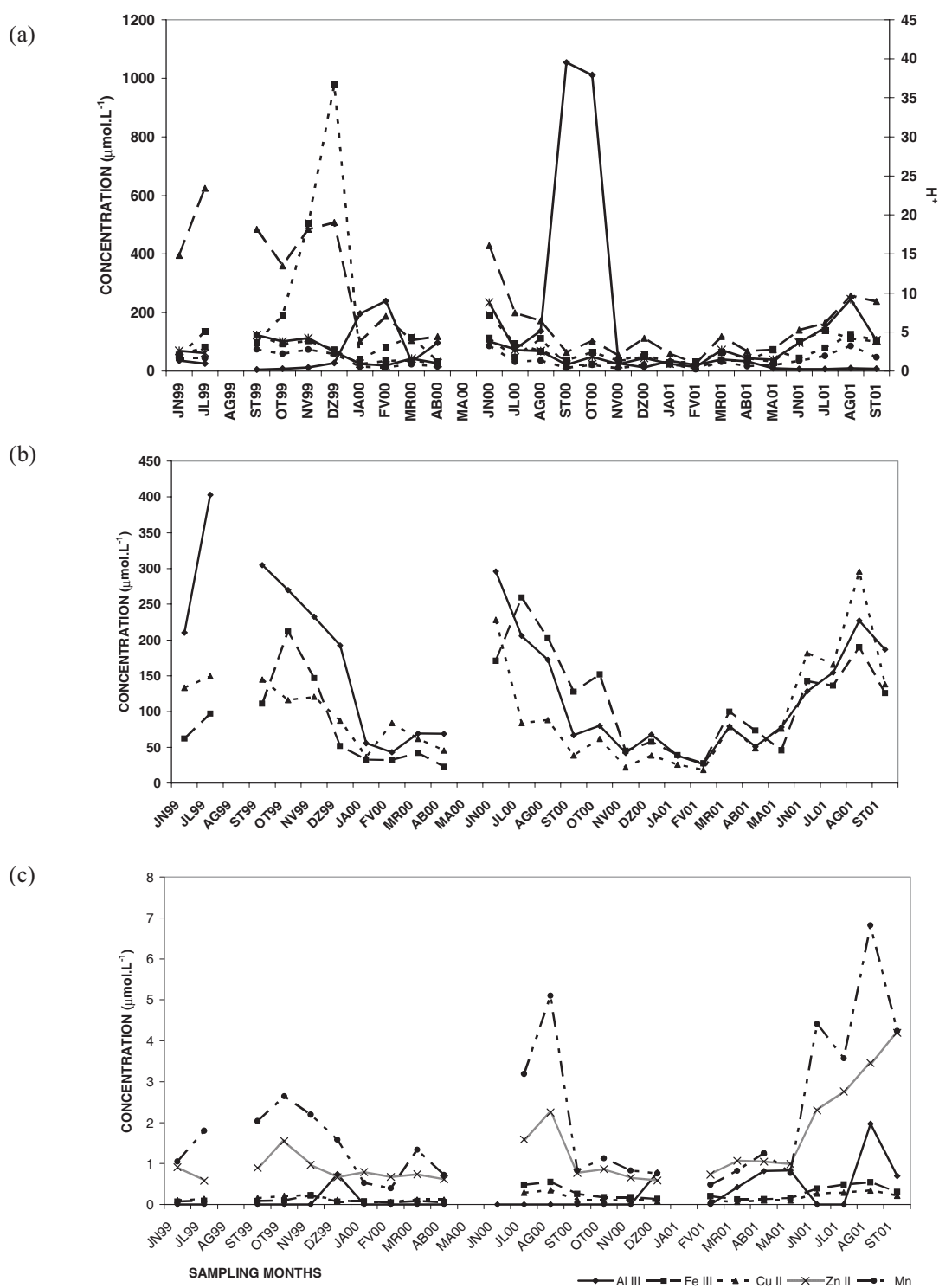


Fig. 5. Throughfall monthly mean concentration variations in different chemicals at PEFI: (a) major cations (b) major anions (c) transition metals.

that become trapped in the canopies. The foliage is an active scavenger for several atmospheric components; when rain falls, it interacts with the components in the canopy to form throughfall solutions which interact with the vegetation. Several processes are involved such as wash of soluble species like Na^+ and Cl^- deposited over the vegetation,

leaching of soluble species present like K^+ in the foliar tissue or like Ca^{2+} in its structure, the absorption of available nutrients such as NH_4^+ and NO_3^- or precipitated like Al III due to pH changes. Even trace metals, essential elements for biological processing, can be leached. In polluted areas such as PEFI, where the anthropogenic inputs are very high,

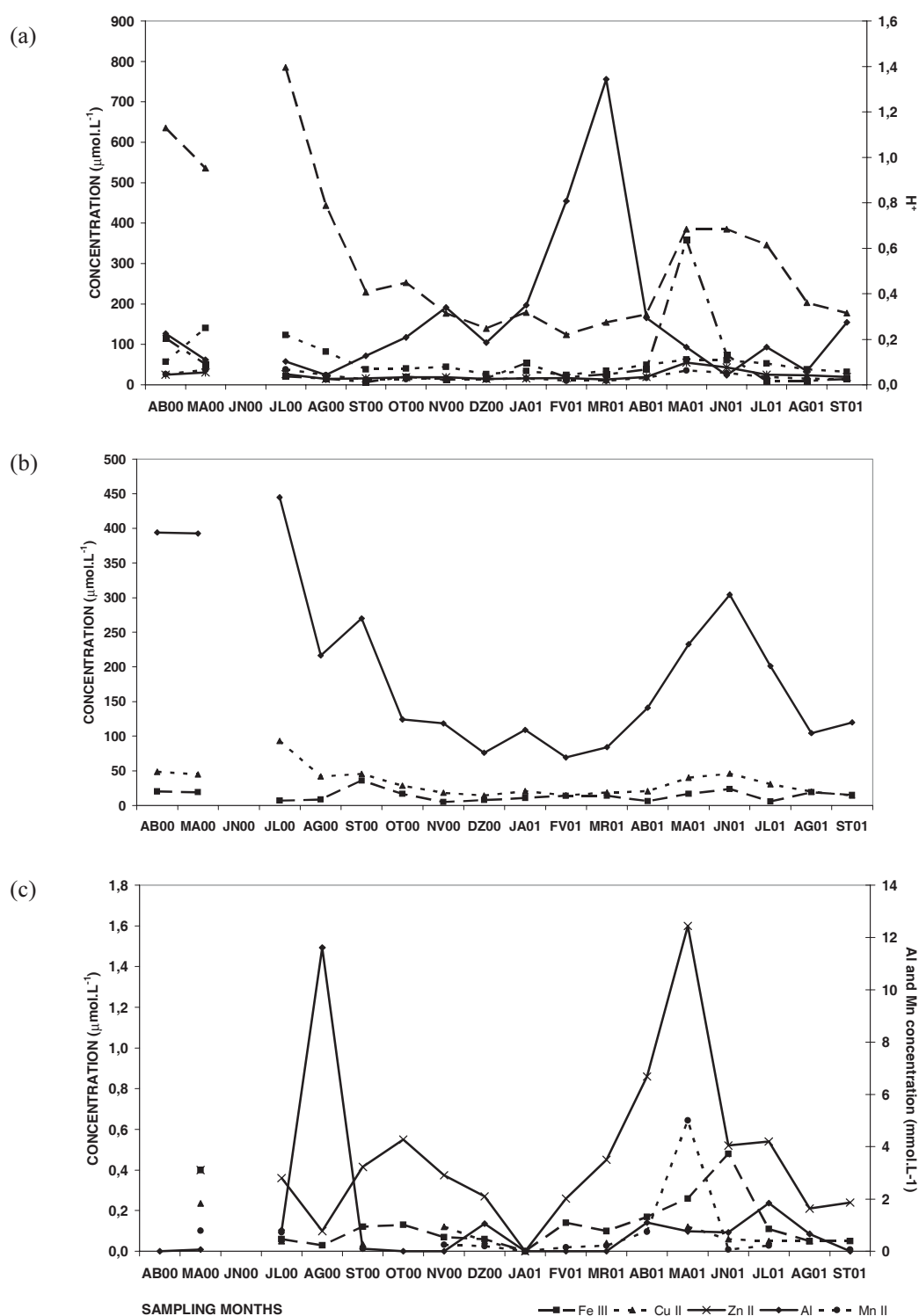


Fig. 6. Throughfall monthly mean concentration variations in different chemicals at CUNHA: (a) major cations (b) major anions (c) transition metals.

the natural processes of throughfall formation are altered due to the excess of chemicals present in the rainfall and deposited on the vegetation.

Table 2 presents the throughfall chemistry for the whole period and for the four driest and wettest months at both PEFI and CUNHA. The mean throughfall pH is higher at

CUNHA (6.4) than at PEFI (5.3) and, although at CUNHA a seasonal difference is significant statistically, it is not so at PEFI. The PEFI's throughfall is slightly acid indicating that the bases leached from the vegetation and solubilised from the dry deposited material are insufficient to neutralise the acidity introduced by the rainfall. A hypothesis test of

Table 2. Throughfall chemistry in $\mu\text{Eq.L}^{-1}$ for the whole (TOTAL) period and the driest and wettest months obtained for PEFI and CUNHA. PTR: throughfall amount mm; N: sample number; MAX: maximum values; MIN: minimum values, MP: volume weighted mean.

		PEFI			CUNHA		
		<i>Total</i>	<i>Driest</i>	<i>Wettest</i>	<i>Total</i>	<i>Driest</i>	<i>Wettest</i>
PTR	(mm)	2726	488.2	1445	2975	498.8	1672
H ⁺	N	62	26	36	50	23	27
	MAX $\mu\text{Eq.L}^{-1}$	62.1	6.23	12.5	1.36	0.20	0.75
	MIN	0.10	0.10	0.40	0.04	0.04	0.13
	MP	5.45	1.96	4.08	0.39	0.12	0.41
Na ⁺	MAX $\mu\text{Eq.L}^{-1}$	135	135	48.6	129	129	32.0
	MIN	11.0	41.1	15.8	19.9	32.7	19.9
	MP	35.2	64.6	28.6	36.4	62.1	27.3
K ⁺	MAX $\mu\text{Eq.L}^{-1}$	518	518	334	782	782	218
	MIN	11.8	84.8	45.5	98.9	167	99.0
	MP	125	224	107	189	361	128
Mg ²⁺	MAX $\mu\text{Eq.L}^{-1}$	114	114	40.3	43.4	43.4	29.6
	MIN	9.57	17.8	9.57	9.41	10.4	9.41
	MP	20.9	37.6	15.7	16.2	20.3	14.1
Ca ²⁺	MAX $\mu\text{Eq.L}^{-1}$	245	245	55.3	46.2	40.4	18.8
	MIN	15.2	32.6	15.2	9.02	9.02	9.90
	MP	30.1	78.8	23.8	17.8	20.3	14.4
NH ₄ ⁺	MAX $\mu\text{Eq.L}^{-1}$	467	193	379	115	76.4	38.4
	MIN	0.97	38.4	0.97	8.01	8.01	8.67
	MP	68.0	66.6	63.0	24.5	27.1	22.6
Cl ⁻	MAX $\mu\text{Eq.L}^{-1}$	366	366	163	442	442	135
	MIN	25.8	61.5	34.9	53.7	98.3	53.7
	MP	74.2	165	48.7	116	228	76.0
NO ₃ ⁻	MAX $\mu\text{Eq.L}^{-1}$	269	269	78.1	39.7	39.7	15.3
	MIN	22.5	56.2	25.7	2.30	6.14	2.30
	MP	65.5	121	39.3	9.85	20.0	6.23
SO ₄ ²⁻	MAX $\mu\text{Eq.L}^{-1}$	296	296	84.7	114	114	22.9
	MIN	21.7	45.3	21.7	7.06	16.6	7.06
	MP	55.5	104	40.6	21.0	46.0	13.4
Al III	N	43	16	27	20	13	7
	MAX $\mu\text{Eq.L}^{-1}$	1.97	1.97	0.78	10.1	10.1	1.23
	MIN	0.01	0.01	0.01	0.00	0.02	0.00
	MP	0.18	0.21	0.12	0.51	2.20	0.22
Fe III	MAX $\mu\text{Eq.L}^{-1}$	0.54	0.54	0.22	0.48	0.48	0.36
	MIN	0.01	0.01	0.05	0.00	0.05	0.00
	MP	0.14	0.25	0.10	0.07	0.13	0.06
Cu II	MAX $\mu\text{Eq.L}^{-1}$	0.34	0.34	0.18	0.25	0.25	0.20
	MIN	0.01	0.01	0.04	0.00	0.03	0.00
	MP	0.10	0.18	0.07	0.04	0.07	0.03
Zn II	MAX $\mu\text{Eq.L}^{-1}$	4.19	3.45	0.98	1.60	1.60	0.34
	MIN	0.01	0.01	0.48	0.00	0.21	0.00
	MP	0.93	1.35	0.56	0.28	0.62	0.15
Mn II	MAX $\mu\text{Eq.L}^{-1}$	6.82	6.82	1.86	5.00	5.00	0.21
	MIN	0.01	0.01	0.29	0.00	0.00	0.00
	MP	1.12	2.20	0.55	0.28	1.15	0.08

the difference between the mean values for throughfall for the wet and dry periods was undertaken for all chemicals at both sites. There was no statistically significant difference for H^+ , K^+ , NH_4^+ and $Al\ III$ for PEFI and for Mg^{2+} , NH_4^+ and NO_3^- for CUNHA. The other ions showed significant differences between wet and dry periods.

PCA analyses were carried out for PEFI and CUNHA rainfall and throughfall variables separately. According to the broken-stick model, PCA extracted two interpretable axes of variation from the 13 chemical water variables (Tables 3 and 4) over the entire observational period accordingly with the climatic characteristics: the wettest, driest and transitional months (time interval between the wet and dry seasons).

For PEFI rainfall (Fig. 7a) and throughfall (Fig. 7b), the eigen-values of the two first principal components accounted for 69.2% and 76.5% of the total variance of the observations, respectively (Table 3). Six (rainfall) and eight (throughfall) variables contributed most to the construction of component 1 ($r > 0.6$). This axis was related negatively to major ionic species except for NH_4^+ and NO_3^- for rainfall (Table 3). On the second axis, transition metals were the main variables indicating the diverse processes involved in the incorporation of transition metals in the rainfall and throughfall. For rainfall chemistry, both axes do not contrast

samples from the climatic characteristics indicating a lack of seasonal variation in agreement with the tests of the hypothesis. However, for throughfall chemistry, the first axis contrasts samples from the dry months (which were positively associated with major ionic species) with those from the other periods, mainly the wet months.

The eigen-values of the two first principal components for CUNHA, rainfall (Fig. 8a) and throughfall (Fig. 8b) represent, respectively, up to 60.7% and 60% of the total variance of the observations (Table 4). For rainfall, five variables contributed most to the construction of component 1 ($r > 0.6$). This axis was negatively related with the major ionic species except for Na^+ and Ca^{2+} (Table 4). On the second axis, two transition metals, (Fe III and Cu II) and Na^+ were the main variables. For throughfall, the major ionic species were highly correlated ($r > 0.6$) with the first axis (except for NO_3^-) and all transition metals to the second component, indicating that the terrigenous source is accountable for a significant part (16%) of the total variance (Table 4). As was found for PEFI, the analysis indicates no seasonal variation in rainfall chemistry. However, throughfall chemistry was even more clearly separated into wet and dry periods than in PEFI.

Table 3. Loadings of variables on the first two principal components for rainfall (R) and throughfall (TR) water samples in PEFI and the proportion of variance explained by each component.

Variable	Rainfall		Throughfall	
	PC 1-R	PC 2-R	PC 1-TR	PC 2-TR
Na^+	-0.78	0.40	-0.83	0.25
K^+	-0.73	0.38	-0.68	0.39
Mg^{2+}	-0.91	0.32	-0.93	0.29
Ca^{2+}	-0.93	0.20	-0.93	0.19
NH_4^+	0.08	0.31	-0.58	0.50
Cl^-	-0.80	0.42	-0.91	0.32
NO_3^-	-0.07	-0.48	-0.78	0.03
SO_4^{2-}	-0.85	0.24	-0.90	0.20
$Al\ III$	-0.29	-0.84	-0.18	-0.42
Fe III	-0.18	-0.80	-0.42	-0.80
Cu II	-0.71	-0.59	-0.70	-0.62
Zn II	-0.63	-0.63	-0.49	-0.81
Mn II	-0.58	-0.61	-0.72	-0.63
Observed eigenvalue	5.56	3.43	6.94	3.04
Broken-Stick eigenvalue	3.18	2.18	3.18	2.18
% of variance	42.80	26.37	53.40	23.11

Table 4. Loadings of variables on the first two principal components for rainfall (R) and throughfall (TR) water samples in CUNHA and the proportion of variance explained by each component.

Variable	Rainfall		Throughfall	
	PC 1-R	PC 2-R	PC 1-TR	PC 2-TR
Na^+	-0.24	0.82	-0.93	-0.13
K^+	-0.82	-0.18	-0.92	-0.32
Mg^{2+}	-0.89	-0.10	-0.84	-0.17
Ca^{2+}	-0.36	-0.44	-0.82	-0.01
NH_4^+	-0.65	-0.40	-0.51	0.31
Cl^-	-0.68	0.49	-0.90	-0.36
NO_3^-	-0.57	0.42	-0.29	-0.44
SO_4^{2-}	-0.87	0.11	-0.88	-0.28
$Al\ III$	—	—	-0.53	0.39
Fe III	-0.60	-0.71	-0.45	0.40
Cu II	-0.18	-0.67	-0.39	0.20
Zn II	-0.40	0.49	-0.37	0.79
Mn II	-0.60	0.24	-0.45	0.76
Observed eigenvalue	4.53	3.10	6.19	3.25
Broken-Stick eigenvalue	2.75	2.10	2.25	2.20
% of variance	37.77	22.90	44.19	15.81

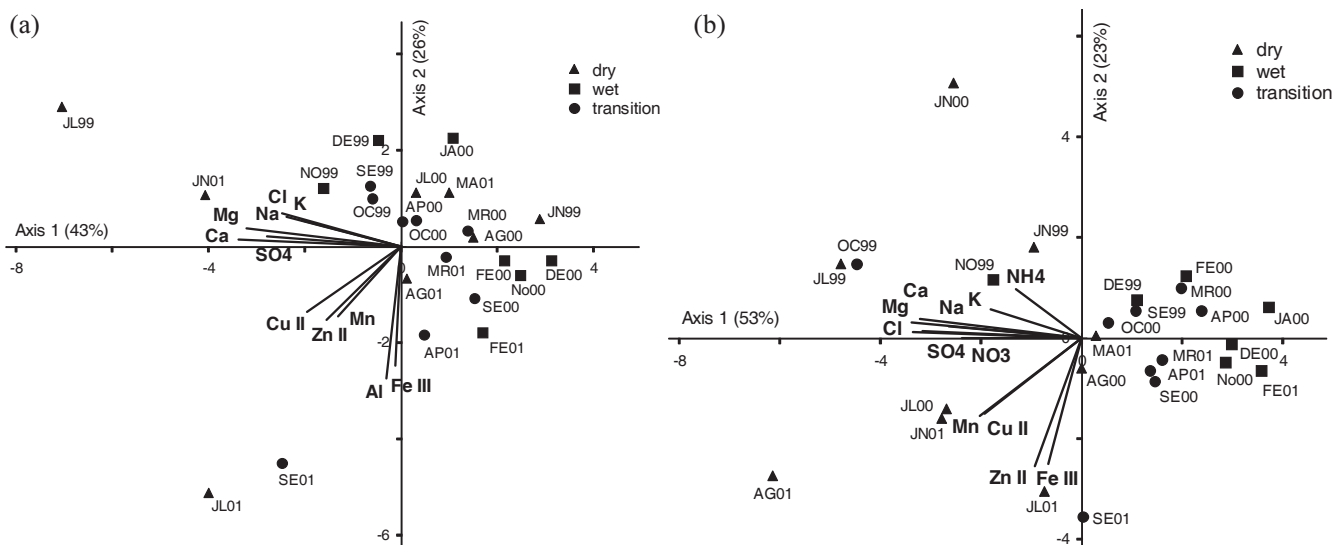


Fig. 7. PCA biplot of variables and scores for the rainfall (a) and throughfall (b) water samples at PEFI. The samples are labelled with month and year, with the symbols representing the climatic periods. For variables correlation with principal components see Table 3.

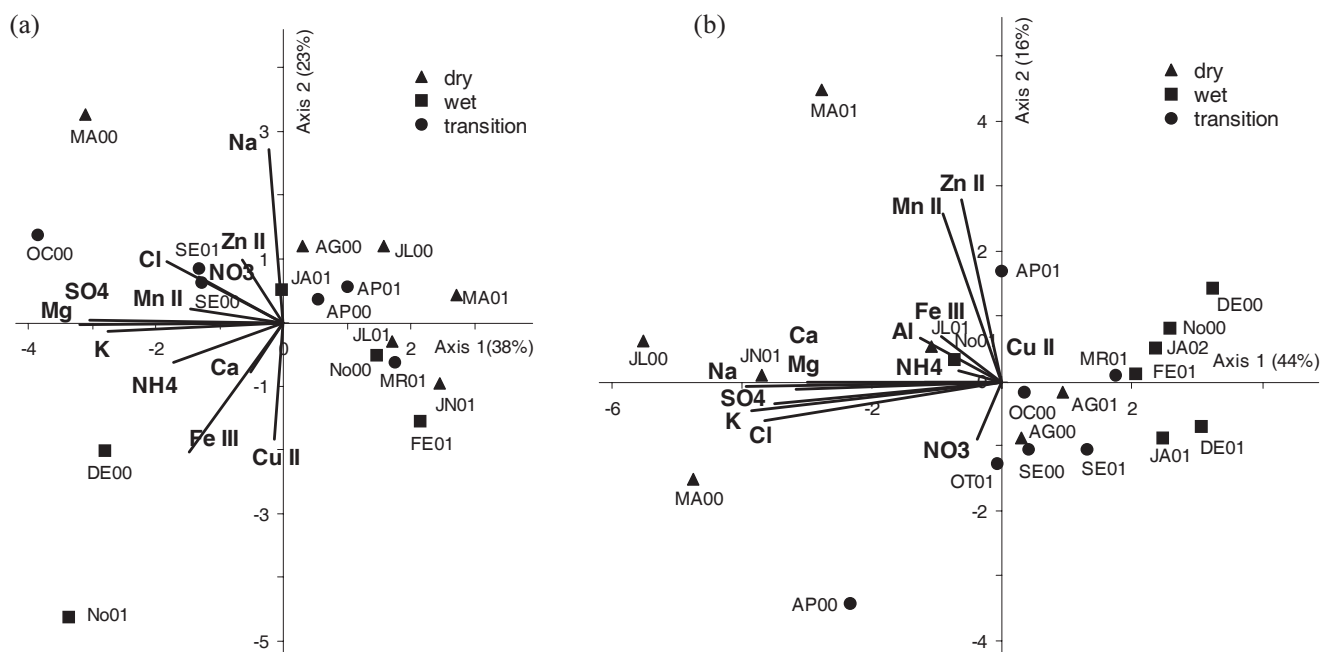


Fig. 8. PCA biplot of variables and scores for the rainfall (a) and throughfall (b) water samples at CUNHA. The samples are labelled with month and year, with the symbols representing the climatic periods. For correlation of variables with principal components see Table 4.

DEPOSITION OF CHEMICAL SPECIES FROM RAINFALL AND THROUGHFALL

The volume weighted mean values of the total amount of each species deposited in $\text{kg ha}^{-1} \text{yr}^{-1}$ were determined by multiplying the concentration by the annual rainfall amount while seasonal deposition was calculated for the four wettest and driest months. (Table 5).

As regards annual depositions, there are marked

anthropogenic inputs of Ca^{2+} , NO_3^- and SO_4^{2-} and transition metals at PEFI as compared to CUNHA, but Al III comes from soil dust sources at CUNHA. The estimated depositions for the transition metals are a fraction of the actual ones because in this work only their cation form is considered. The NH_4^+ fluxes are equivalent in the two areas during the rainy period; however, equivalent depositions do not mean equivalent sources. During the wettest period larger

Table 5. Annual ($\text{kg} \cdot \text{ha}^{-1} \cdot \text{yr}^{-1}$) and seasonal deposition ($\text{kg} \cdot \text{ha}^{-1}$: four wettest and driest months) through the rainwater (CH) and throughfall (TR) for PEFI and CUNHA areas.

	PEFI						CUNHA					
	Year		Driest		Wettest		Year		Driest		Wettest	
	CH	TR	CH	TR	CH	TR	CH	TR	CH	TR	CH	TR
	$\text{kg ha}^{-1} \text{yr}^{-1}$		kg ha^{-1}		kg ha^{-1}		$\text{kg ha}^{-1} \text{yr}^{-1}$		kg ha^{-1}		kg ha^{-1}	
H ⁺	0.59	0.06	0.02	0.01	0.32	0.03	0.08	0.01	0.01	0.00	0.04	0.00
Na ⁺	4.49	8.82	1.89	3.63	2.45	4.75	3.56	12.4	0.96	3.56	1.20	5.26
K ⁺	5.40	53.1	1.26	21.4	4.65	30.4	6.26	110	0.31	35.2	3.80	42.0
Mg ²⁺	1.34	5.55	0.45	2.23	0.76	2.75	1.04	5.86	0.09	1.23	0.62	2.87
Ca ²⁺	7.41	16.7	2.55	7.72	3.94	6.89	1.41	10.6	0.16	2.03	0.81	4.84
NH ₄ ⁺	13.6	13.4	1.85	2.93	9.48	8.19	12.7	6.56	0.53	1.22	10.4	3.41
Cl ⁻	12.1	28.7	4.79	14.3	7.29	12.5	9.53	61.3	1.29	20.2	2.92	22.6
NO ₃ ⁻	50.8	44.3	18.1	18.4	28.0	17.6	9.23	9.09	2.01	3.09	3.44	3.23
SO ₄ ²⁻	28.6	58.1	7.33	24.4	18.9	28.2	7.43	30.1	1.22	11.0	3.56	10.7
Al III	0.06	0.05	0.01	0.01	0.02	0.02	1.47	0.20	0.14	0.15	1.01	0.05
Fe III	0.19	0.09	0.04	0.03	0.11	0.04	0.04	0.06	0.01	0.02	0.03	0.03
Cu II	0.07	0.07	0.02	0.03	0.04	0.03	0.07	0.03	0.01	0.01	0.06	0.02
Zn II	1.79	0.67	0.46	0.22	0.60	0.26	0.29	0.27	0.08	0.10	0.15	0.08
Mn II	0.13	0.67	0.05	0.29	0.06	0.22	0.04	0.23	0.01	0.16	0.02	0.04

quantities of all species are deposited than in the driest one because during the wettest months, higher rainfall results in higher depositions at both sites. These results are higher than those from other Brazilian cities (Forti *et al.*, 2001) (Table 6). Depositions at CUNHA are similar to those in a town in the same region on the sea border (Caraguatatuba); values are higher than those found in the Amazon forest for nitrate and sulphate. However, the amount of nitrate at PEFI is by far the largest of all cities.

Throughfall net depositions differ in the two areas, reflecting not only the polluted atmosphere at PEFI and the relatively clean one at CUNHA but also differences in the vegetation and in the nutritional status of that vegetation; these values are shown in Table 7. NH₄⁺ and NO₃⁻ are both absorbed in a dry period but less NO₃⁻ is absorbed in the wet one, when NH₄⁺ is removed from the vegetation.

The deposition rates for Na⁺, K⁺, Mg²⁺ and Ca²⁺ at CUNHA are similar for the two periods studied. NH₄⁺ is absorbed in both, in larger amounts in the wettest period and NO₃⁻ is absorbed in the driest period. In the wettest period, NO₃⁻ is deposited mainly in rainfall. More Cl⁻ and SO₄²⁻ are deposited in the dry period, mainly in throughfall: the main source of these species is the sea and they are deposited as gas or particles on the vegetation from which they are subsequently washed by the rain.

The concentrations for the transition metals may be underestimated because they refer only to their free metallic ions form rather than to the total amount and transition

Table 6. Nitrate and sulphate deposition values (in $\text{kg ha}^{-1} \text{yr}^{-1}$) for other Brazilian regions

Areas near sea shore	NO ₃ ⁻	SO ₄ ²⁻
<u>Highly-Moderately industrialised (more than a million inhabitants)</u>		
Porto Alegre	6.82	178
Florianópolis	4.96	268
Cubatão/Mogi	48.4	310
Fortaleza	4.96	13.5
Niterói/RJ	8.06	22.6
<u>Urban (less than 500 million inhabitants)</u>		
Caraguatatuba	8.06	21.1
Natal	0.62	3.36
<u>Rural</u>		
Cubatão/Paranapiacaba	5.58	44.2
Piracicaba*	9.92	12.0
<u>Amazon Forest</u>		
Central**	4.96	20.1
Northeastern	4.34	14.9

Source: Forti *et al.* 2001

* more than 150 km from sea

** about 1000 km from sea

Table 7. Net throughfall deposition ($\text{kg}\cdot\text{ha}^{-1}\cdot\text{yr}^{-1}$) for the whole period and the driest and wettest ones ($\text{kg}\cdot\text{ha}^{-1}$).

	PEFI			CUNHA		
	Total $\text{kg}\cdot\text{ha}^{-1}\cdot\text{yr}^{-1}$	Driest $\text{kg}\cdot\text{ha}^{-1}$	Wettest $\text{kg}\cdot\text{ha}^{-1}$	Total $\text{kg}\cdot\text{ha}^{-1}\cdot\text{yr}^{-1}$	Driest $\text{kg}\cdot\text{ha}^{-1}$	Wettest $\text{kg}\cdot\text{ha}^{-1}$
H^+	-0.53	-0.01	-0.29	-0.07	-0.01	-0.04
Na^+	4.33	1.74	2.30	8.84	2.60	4.06
K^+	47.7	20.1	25.8	104	34.9	38.2
Mg^{2+}	4.21	1.78	1.99	4.82	1.14	2.25
Ca^{2+}	9.29	5.17	2.95	9.19	1.87	4.03
NH_4^+	-0.20	1.08	-1.29	-6.14	0.69	-6.99
Cl^-	16.6	9.51	5.21	51.8	18.9	19.7
NO_3^-	-6.50	0.30	-10.4	-0.14	1.08	-0.21
SO_4^{2-}	29.5	17.1	9.30	22.7	9.78	7.14
Al III	-0.01	0.00	0.00	-1.27	0.01	-0.96
Fe III	-0.10	-0.01	-0.07	0.02	0.01	0.00
Cu II	0.00	0.01	-0.01	-0.04	0.00	-0.04
Zn II	-1.12	-0.24	-0.34	-0.02	0.02	-0.07
Mn II	0.54	0.24	0.16	0.19	0.15	0.02

metals in colloidal and complexed form can be important. Therefore, their estimated deposition rates may well represent only a fraction of the real input to the areas. The natural source of Al III is the soil, though in PEFI, the Al III anthropogenic contribution is important and Al III may be solubilised from the dust deposited over the canopy. As a function of changes in the pH of the solution, deposition values may indicate that precipitation of Al III may be occurring. The same is probably true for Fe III, while Zn II and Mn II are important micronutrients and therefore active species within the canopy. The deposition values indicate that the vegetation is absorbing Zn II and that Mn II is being leached from the vegetation. Cu II is found in trace amounts. Studies of the chemical composition of urban particulate matter showed that trace metals such as Cu, Mn and Zn are present in readily water-soluble and available forms (Fernández-Espinosa *et al.*, 2002; Voutsas and Samara, 2002). Kyotani and Iwatsuki (2002) pointed out that most of these elements are present as sulphates, nitrates and chlorides; this explains why they are leached easily from vegetation by rain and throughfall after dry deposition and/or rain events. Knowledge of these transition metals in rainfall and throughfall is rare and detailed studies are necessary to understand their functioning in these solutions.

The removal of active elements in the biogeochemical cycles as K, Mn and N reflects leaching or consumption processes by the foliar surfaces. K^+ is easily leached from the vegetation and its flux at CUNHA is twice that calculated for PEFI. Arcova and Cicco (1987) also observed high K

fluxes in this region. In other tropical forest regions such as the preserved forests of Panama (Golley *et al.*, 1978), in São Paulo State, (Meguro *et al.*, 1979), and in the Cardoso Island Atlantic forest (Varjabedian, 1994), amongst others, high potassium fluxes in throughfall have been found. According to Lopes (2000), the high humidity environment of these tropical forests favours K internal recycling as well as its leaching from the arboreal vegetation and its associated epiphytic flora. The lower fluxes of K^+ observed at PEFI can be attributed to a deficiency of that nutrient in the vegetation, which can be related to a lack of K^+ in the soil. Fluxes of Mg^{2+} in PEFI are larger than in CUNHA, reflecting a larger availability in the vegetation; its leaching from the vegetation has been associated with low concentrations in the soil solution and in the drainage waters so that Mg is recycled quickly by the vegetation (Nihlgård *et al.*, 1994).

Mg^{2+} is an important nutrient and is usually leached from the vegetation together with K^+ . Although an anthropogenic source of Mg certainly exists at PEFI, its flux is the same as that at CUNHA, suggesting that the vegetation at PEFI is also deficient in Mg. The smaller fluxes of K^+ and Mg^{2+} at PEFI may well be a consequence of the high deposition rates of anions that take with them the basic cations. One of the effects related strongly with soil acidification is the leaching of nutrients such as potassium, calcium, magnesium and ammonium (Lopes, 2000).

In CUNHA, deposition rates are low. NH_4^+ absorption occurred mainly during the rainy period, at which season the transfer of NO_3^- to throughfall was low. In PEFI, deposition rates are high with no substantial transference of both ions to throughfall. NH_4^+ is absorbed in the dry season and transferred at low rates during the rainy season, whereas NO_3^- is absorbed at high rates over the whole year. Annual figures of 1.5% and 12% of NO_3^- absorption were registered for CUNHA and PEFI, respectively.

Such differences in deposition, absorption, and transference of NH_4^+ and NO_3^- probably reflect the composition and functioning of the natural plant communities in the contrasting areas. In the heavily polluted industrial area of Cubatão (SP, Brazil), $\text{NH}_3/\text{NH}_4^+$ was the predominant form of nitrogen in wet deposition, as well as a probable nitrogen source for vegetation; the study by Steward *et al.*, (2002) of nitrogen isotope signatures of vegetation in the area indicated the presence of species that used deposited $\text{NH}_3/\text{NH}_4^+$ as their main source of nitrogen, as well as species with a high potential for using NO_2 or NO_3^- as nitrogen sources. Moreover, the vegetative succession must be taken into account as PEFI is a disturbed area. Recent results for an Atlantic Forest succession in southeastern Brazil (Aidar *et al.*, 2003) indicated a continuum of species with different nitrogen use strategies

across the forest. Representing the extremes, the pioneer species had high leaf nitrogen contents and a high capacity for leaf nitrate assimilation, and responded to the increase in soil nitrogen availability, whereas late secondary species had low leaf nitrogen contents, low leaf nitrate assimilation capacity and little capacity to respond to soil nitrogen availability among other characteristics. Therefore, nitrogen transference through the canopy is closely linked to the nitrogen use strategies of different species and may be indicative of anthropogenic sources of nitrogen as well as of the vegetation structure and functioning. PEFI vegetation is a mosaic, with degraded and recovering areas having different successional species and just a few dominant ones (Pivello and Peccinini, 2002). Consequently, the area is expected to have a high capacity for nitrogen use.

Ca^{2+} is more vulnerable to the leaching processes during senescence as it is mostly present in the structural parts of the vegetation. However, such periods were not identified in this work. Some of the Ca^{2+} fluxes at PEFI certainly originated in building construction and in dust resuspended by vehicular traffic.

The presence in the throughfall solution of less active elements in the biogeochemical cycle, such as Cl and S, for which the leaching and consumption are irrelevant, originated from direct deposition over the canopy. These chemicals have a natural source in marine aerosols and in the condensation nuclei in clouds and are incorporated by the throughfall through the washout of the vegetation. At CUNHA, one important source of SO_4^{2-} is the marine aerosol, while at PEFI anthropogenic sources are dominant over the natural ones.

Fluxes of transition metals are generally low even at PEFI and this is certainly because they have been underestimated here. Significant fluxes were found for Zn II and for Mn II, essential metals (micronutrients) for the vegetation; retention of Zn II by the vegetation indicates that it is being fixed. For Mn II, removal rates from the vegetation at PEFI are 80% while at CUNHA they range from 82% on average to 94% during the wettest period.

Conclusions

The characteristics of deposition of the major species and transition metals in contrasting Atlantic Forest areas differ. In an area of remnant forest in an urban environment with high atmospheric pollution, the deposition of solutes from rain is much higher than that far from major urban centres. The analysis of net throughfall fluxes provides evidence for differences in the nutritional status of the vegetation in the two areas. These differences can be attributed to the action of pollutants on the mass of vegetation as well as on

the soils which are affected by high inputs of nitrate and sulphate that lower values of pH, so inducing acid substances in the soil. In addition to the direct effect on the canopy, the local vegetation will become degraded because of a decrease in the nutrient supply and/or a nutritional disequilibrium. As reported by Lopes (2000) for the Atlantic Forest in the Cubatão area, the degradation of vegetation results in a loss of nutrients because, together with the soil organisms, they are important in nutrient storage and recycling and in maintaining the closed internal cycles and preventing losses by leaching.

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